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Sherman et al.

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(54) **METHOD OF AND APPARATUS FOR MANUFACTURING METHANOL**
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(*) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 0 days.

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(List continued on next page.)

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(52) **U.S. Cl.** **204/157.9**
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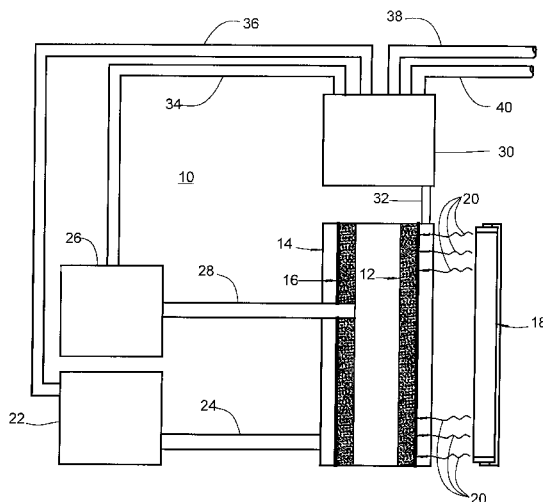
ABSTRACT

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In a method of and apparatus for manufacturing methanol and higher alcohols from natural gas a catalytic area is formed on the exterior of a gas permeable partition. Natural gas is maintained on the interior of the gas permeable partition at predetermined pressure. Relative movement between the gas permeable partition and the water forms sub-micron sized bubbles of natural gas. Electromagnetic radiation is directed onto the catalytic surface to form hydroxyl radicals from the water. Methyl, ethyl, and propyl ions from the natural gas combine with the hydroxyl ions to form methanol, ethanol, and propanol.

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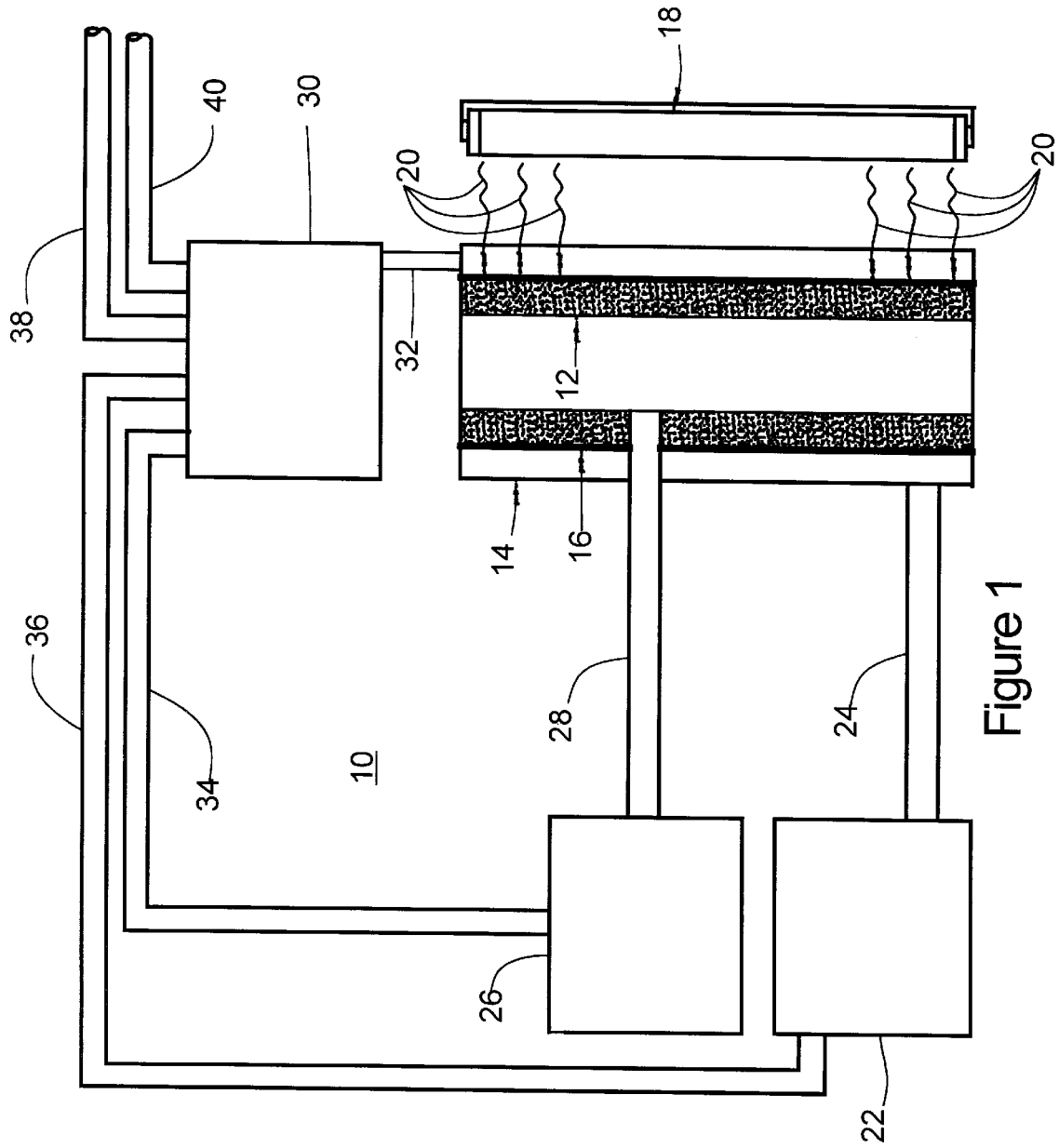


Figure 1

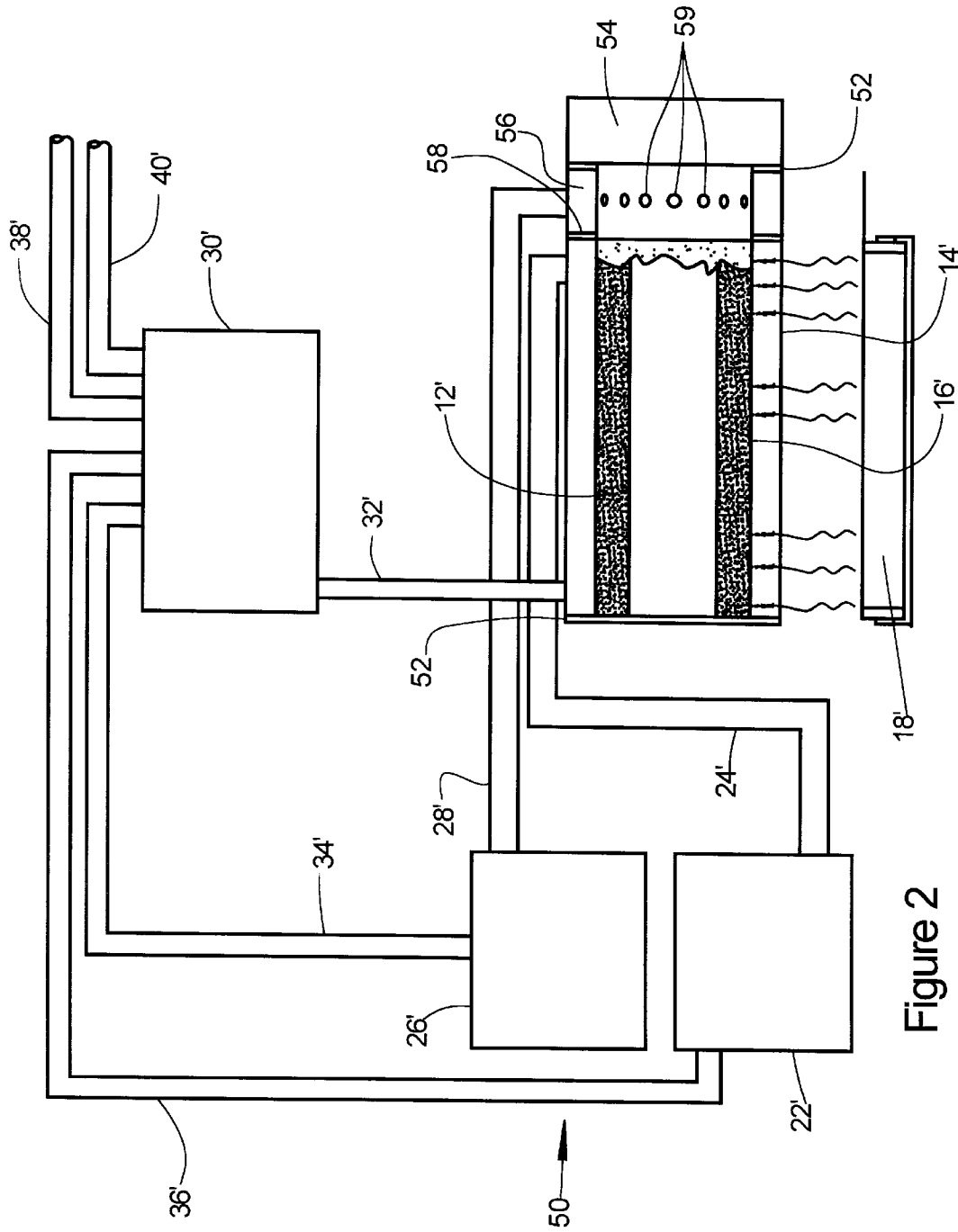


Figure 2

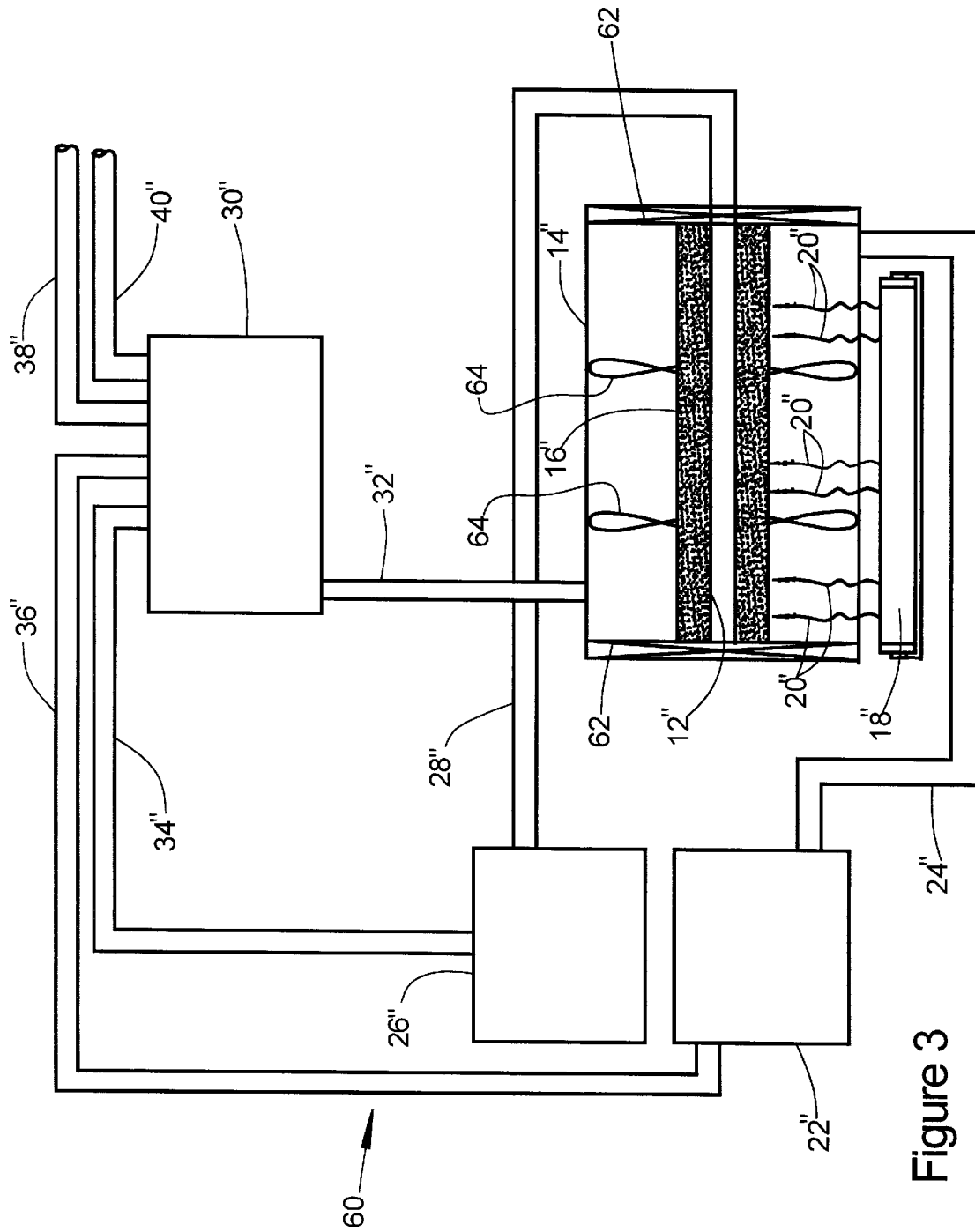


Figure 3

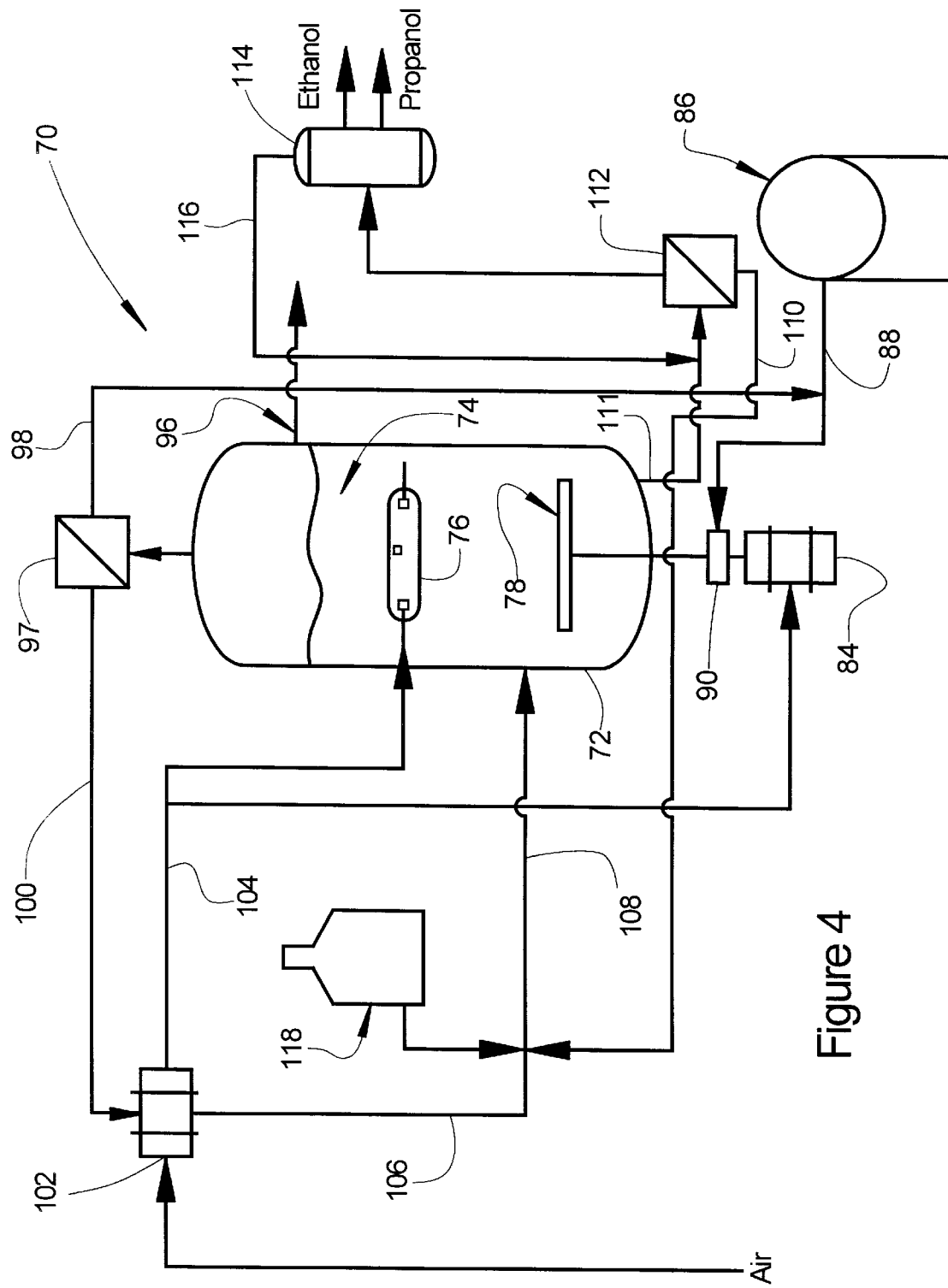


Figure 4

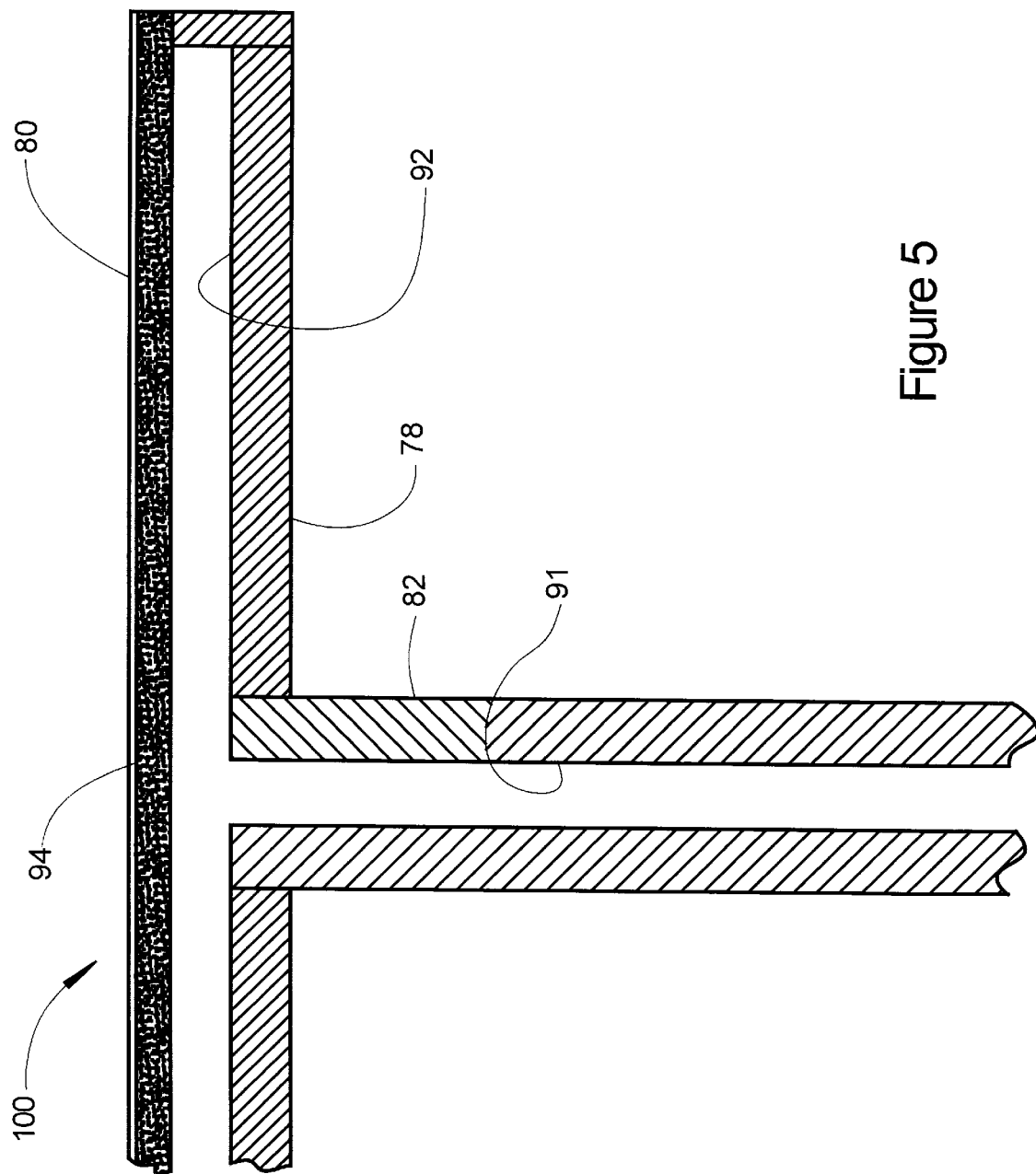


Figure 5

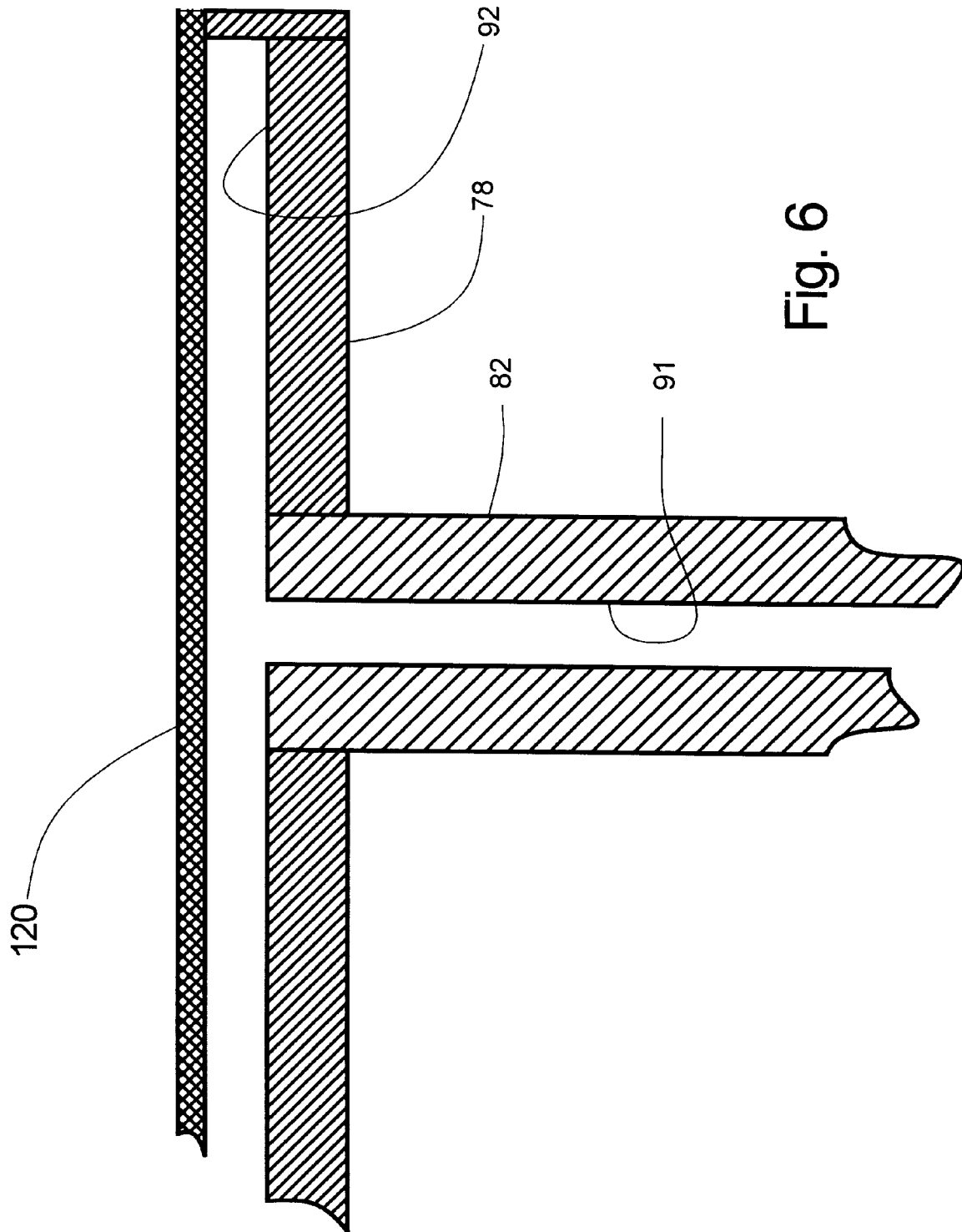


Fig. 6

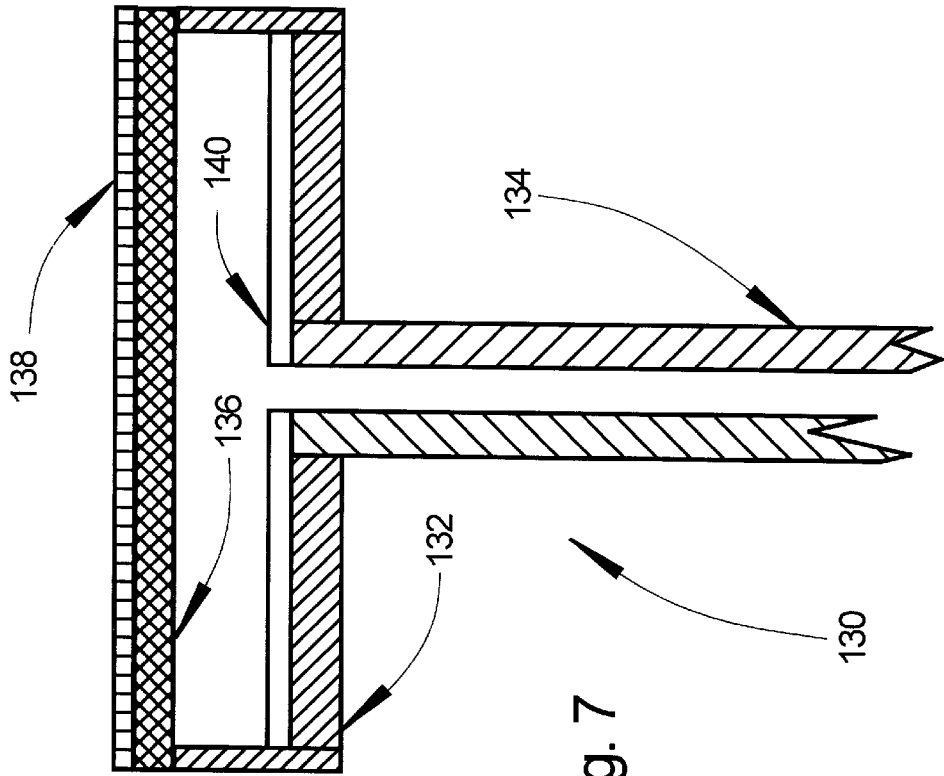
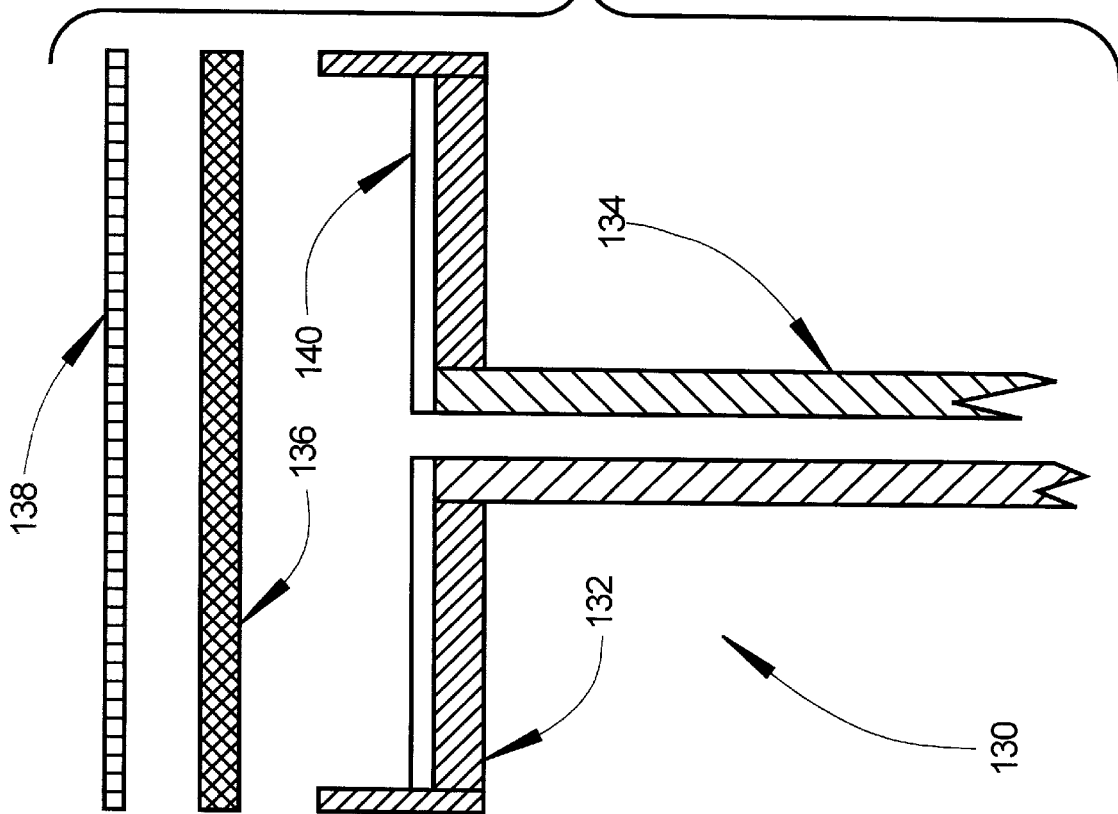


Fig. 7

Figure 8



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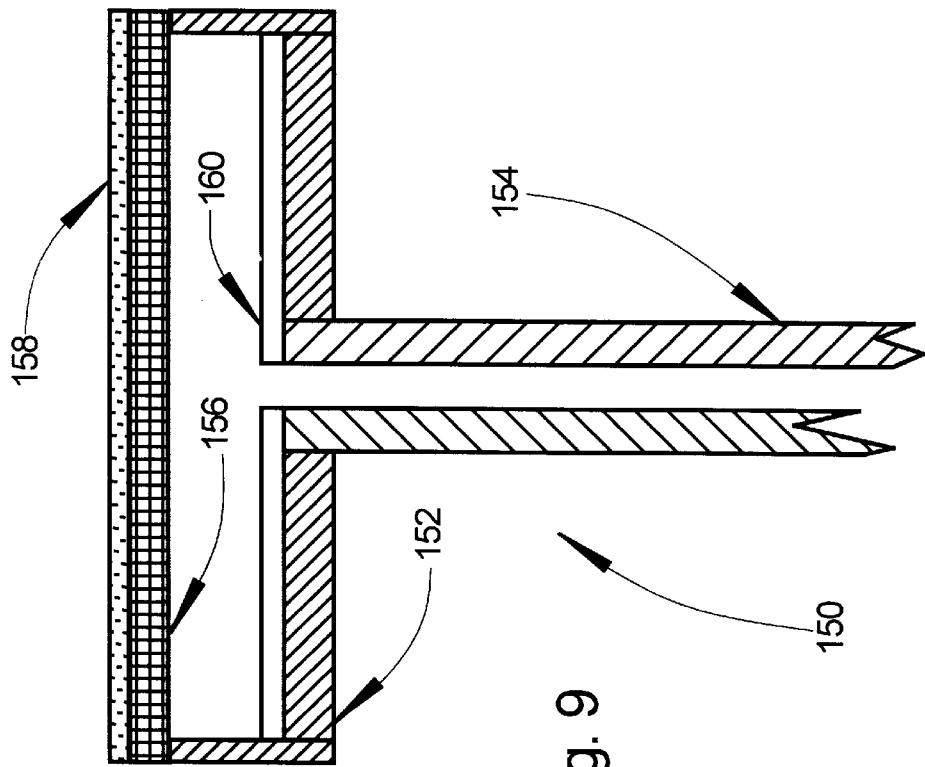


Fig. 9

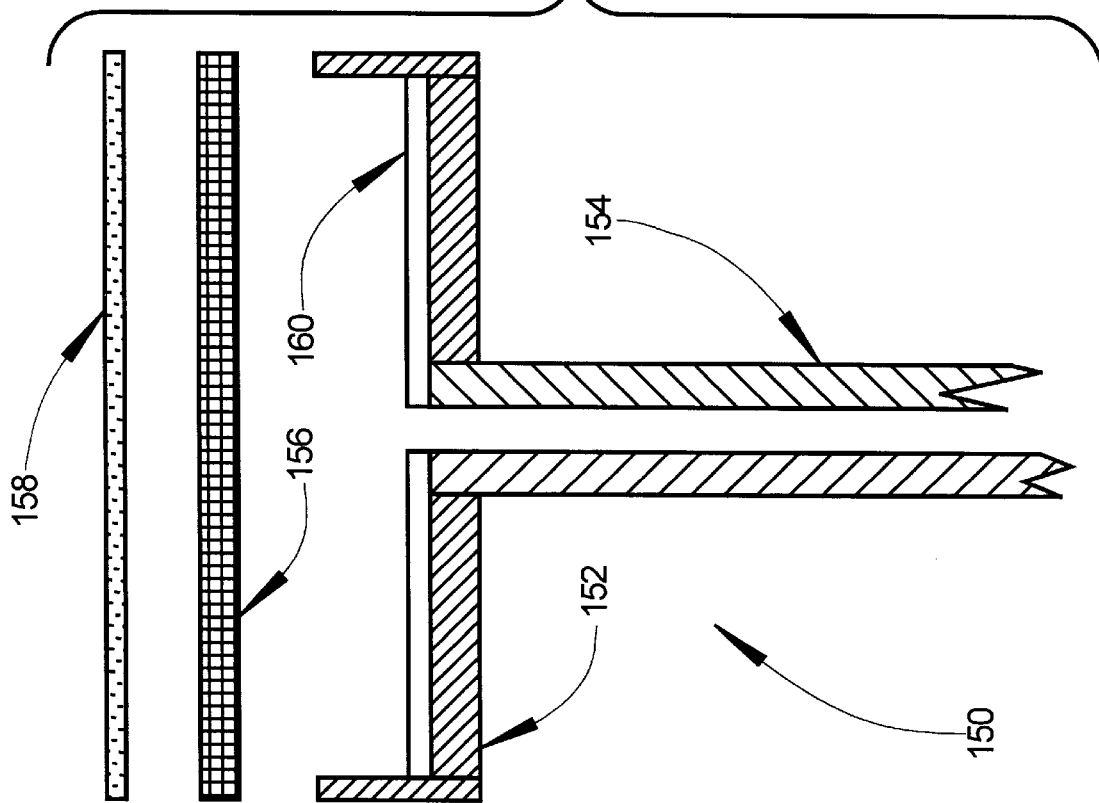
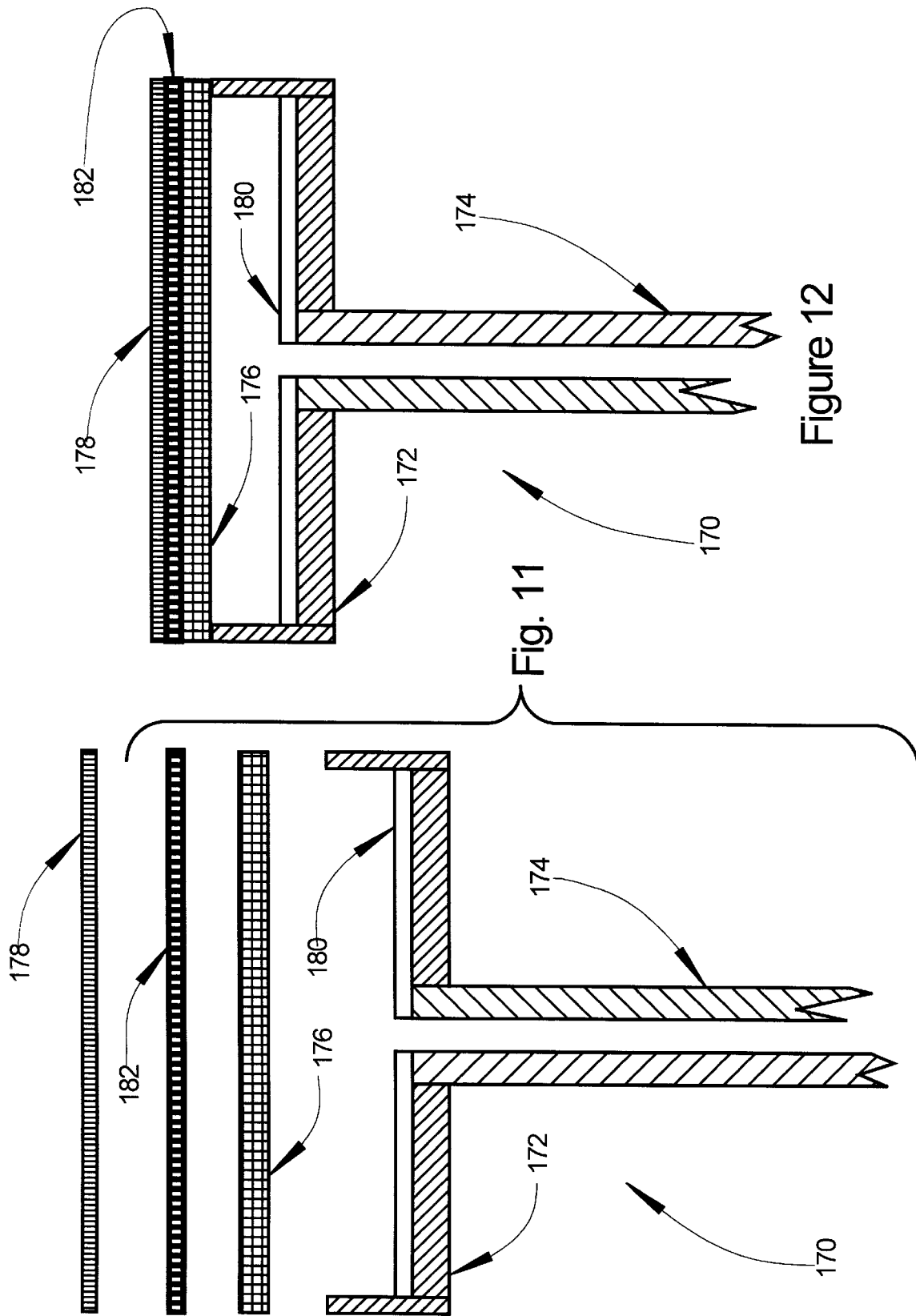


Figure 10



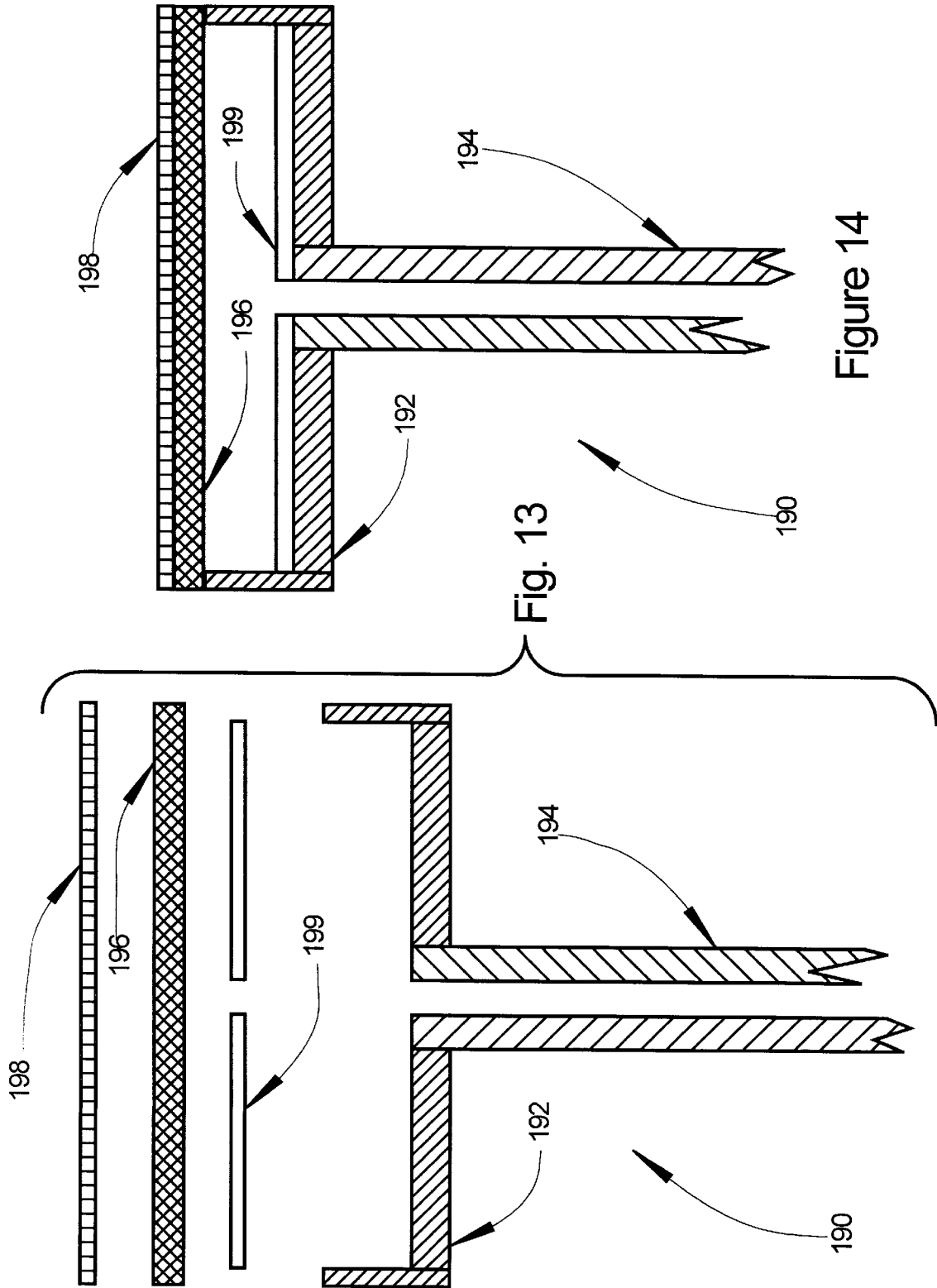


Fig. 13

Figure 14

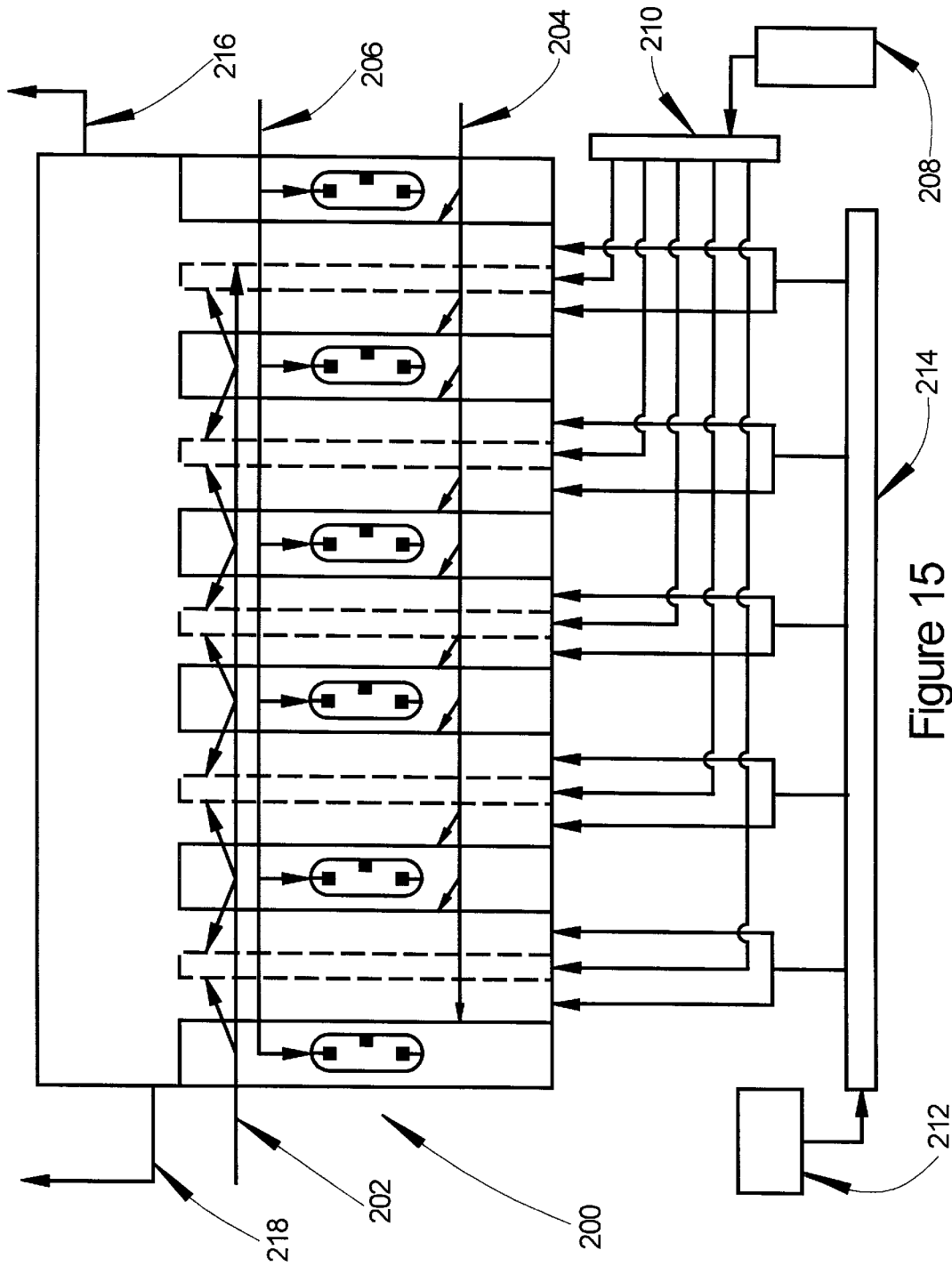


Figure 15

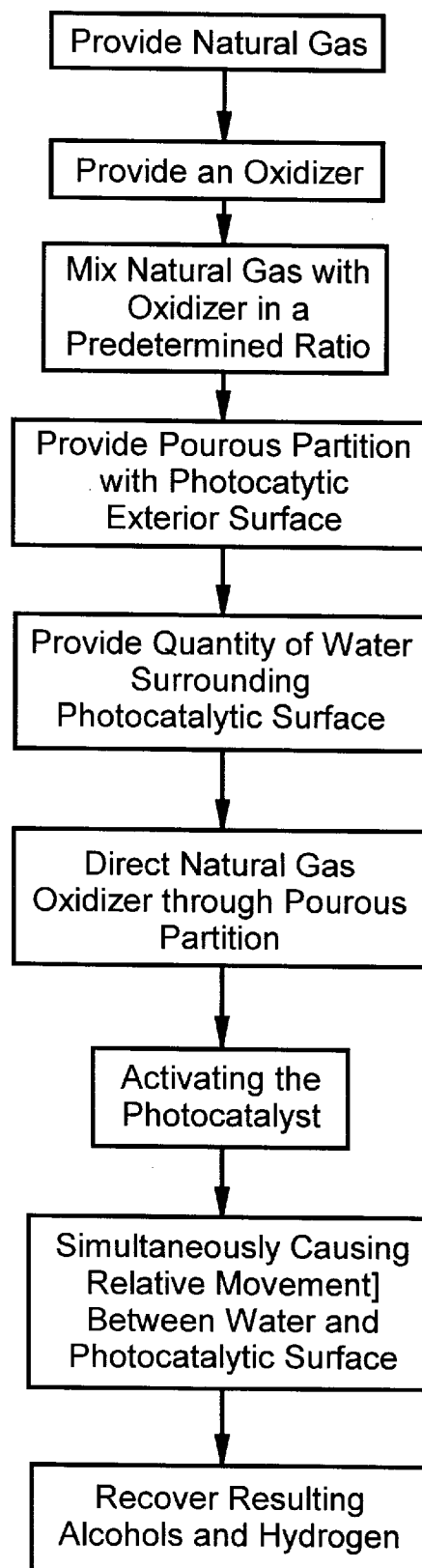


Figure 16

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**METHOD OF AND APPARATUS FOR
MANUFACTURING METHANOL****CROSS REFERENCE TO RELATED
APPLICATIONS**

This application is a continuation-in-part of prior application Ser. No. 09/368,404 filed Aug. 4, 1999, currently pending which is a continuation-in-part of prior application Ser. No. 09/224,394 filed Dec. 31, 1998, now U.S. Pat. No. 6,129,818 which is a continuation-in-part of prior application Ser. No. 09/058,494, filed Apr. 10, 1998, now U.S. Pat. No. 5,954,925.

TECHNICAL FIELD

This invention relates generally to the manufacture of methanol, and more particularly to a method of and apparatus for manufacturing methanol from methane, and to a method of and apparatus for manufacturing methanol, ethanol, and propanol from natural gas.

**BACKGROUND AND SUMMARY OF THE
INVENTION**

Methanol, the simplest of the alcohols, is a highly desirable substance which is useful as a fuel, as a solvent, and as a feedstock in the manufacture of more complex hydrocarbons. In accordance with the method of methanol manufacture that is currently practiced in the petroleum industry, methane is first converted to synthesis gas, a mixture of carbon monoxide and hydrogen. The synthesis gas is then converted over an alumina-based catalyst to methanol. The formation of synthesis gas from methane is an expensive process.

Although often identified as methane, the feedstock for the foregoing synthesis gas process is typically natural gas. As is well known, natural gas often contains significant percentages of sulphur. Since sulphur poisons the catalyst required for its operation, the synthesis gas process for making methanol is further limited by the scarcity of low sulphur natural gas.

As will be apparent, methane and methanol are closely related chemically. Methane comprises a major component of natural gas and is therefore readily available. Despite the advantages inherent in producing methanol directly from methane, no commercially viable system for doing so has heretofore been developed.

The present invention comprises a method of and apparatus for manufacturing methanol from methane or natural gas which overcomes the foregoing and other deficiencies which have long since characterized the prior art. The method involves a gas permeable partition upon which a light-activated catalyst capable of producing hydroxyl radicals from water is deposited, it being understood that as used herein the term "light-activated catalyst" means any catalyst that is activated by electromagnetic radiation regardless of wave length.

Water is present on the catalyst side of the partition and methane or natural gas at positive pressure is present on the opposite side of the partition. The catalyst is exposed to radiation while relative movement is effected between the water and the partition. The radiation-exposed catalyst reacts with the water molecules to form hydroxyl radicals. The gas is forced through the semipermeable partition forming small bubbles in the water. The hydroxyl radicals in the water then undergo a free-radical reaction with the methane in the water to form methanol, and if natural gas is used in the process, ethanol and propanol.

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In accordance with the broader aspects of the invention there is generated a stream of sub-micron sized gas bubbles. Due to their extremely small size, the gas bubbles present an extremely large surface area which increases reaction efficiency. Smaller pores in the gas permeable partition facilitate the formation of smaller bubbles. Additionally, higher relative velocity across the partition surface aids in shearing the bubbles off the surface while they are still small.

In accordance with first, second, and third embodiments of the invention, a gas permeable tube has an exterior coating comprising a titanium-based catalyst. The gas permeable tube is positioned within a glass tube and water is caused to continuously flow through the annular space between the two tubes. Methane or natural gas is directed into the interior of the gas permeable tube and is maintained at a pressure high enough to cause gas to pass into the water and prevent the flow of water into the interior of the gas permeable tube. As the water passes over the gas permeable tube, gas bubbles are continually sheared off of its surface. The gas bubbles thus generated are sub-micron in size and therefore present an extremely large surface area.

Electromagnetic radiation generated, for example, by ultraviolet lamps is directed through the glass tube and engages the titanium-based catalyst to generate hydroxyl radicals in the flowing water. The hydroxyl radicals undergo a free-radical reaction with the methane forming methanol, among other free-radical reaction products. Subsequently, the methanol and other products are separated from the reaction mixture by distillation.

In accordance with fourth, fifth, sixth, seventh, and eighth embodiments of the invention, there is provided a hollow disk which supports a gas permeable partition having an exterior coating comprising a titanium-based catalyst. The disk is positioned within a water filled container. Methane or natural gas is directed into the interior of the disk and is maintained at a pressure high enough to cause gas to pass outwardly through the partition and into the water and to prevent the flow of water into the interior of the disk. The disk and the partition are moved at high speed relative to the water. As the gas permeable partition moves relative to the water, gas bubbles are continually sheared off of its surface. The gas bubbles thus generated are sub-micron in size and then therefore present an extremely large surface area.

Electromagnetic radiation generated, for example, by ultraviolet lamps within the container engages the titanium-based catalyst to generate hydroxyl radicals in the water. The hydroxyl radicals undergo a free-radical reaction with the methane forming methanol, and, if natural gas is used in the process, ethanol and propanol. Subsequently, the methanol and other reaction products are separated from the reaction mixture by distillation.

In the practice of the fifth, sixth, seventh, and eighth embodiments of the invention, utilization of the energy comprising the electromagnetic radiation is maximized by providing a mirror within the hollow disk to reflect electromagnetic radiation passing through the porous partition back to the catalytic material. The mirror may comprise either a mirrored surface of the hollow disk or a separate mirror plate. Fluorescent material is utilized to convert broad-band electromagnetic radiation to radiation having a band width which is specific to the selected catalyst. The fluorescent material may be combined with the porous partition, or with the catalytic layer, or may comprise a distinct layer.

In accordance with a ninth embodiment of the invention, a plurality of parallel porous partitions each having a photocatalytic layer on its exterior surface are mounted in an

array. The array further comprises sources of electromagnetic radiation positioned between each of the tubular porous partition/photocatalytic layer assemblies. Methane or natural gas from a first manifold is directed into the interior of each of the parallel porous partitions. Water from a second manifold is directed across the surface of the photocatalytic layers in the manner of the first three embodiments of the invention. In addition to activating the photocatalytic layers, energy from the electromagnetic radiation sources generally provides sufficient heat to distill the resulting methanol and higher alcohols from the water.

In accordance with a tenth embodiment of the invention, an oxidizer such as oxygen, peroxide, etc. is mixed with methane or natural gas. The mixture is then directed through a porous partition having a photocatalytic layer on its exterior surface. Water is continuously directed across the exterior surface of the porous partition in the manner of the first nine embodiments of the invention. In this manner the reaction is rendered self-sustaining.

BRIEF DESCRIPTION OF THE DRAWINGS

A more complete understanding of the invention may be had by reference to the following Detailed Description when taken in conjunction with the accompanying Drawings wherein:

FIG. 1 is a diagrammatic illustration of a method and apparatus for manufacturing methanol comprising a first embodiment of the present invention.

FIG. 2 is a diagrammatic illustration of a second embodiment of the apparatus of the present invention with a rotating sintered stainless steel tube.

FIG. 3 is a diagrammatic illustration of a third embodiment of the apparatus of the present invention with a rotating sintered stainless steel tube with turbines.

FIG. 4 is a diagrammatic illustration of a fourth embodiment of the apparatus of the present invention.

FIG. 5 is an enlargement of a portion of FIG. 4.

FIG. 6 is an illustration similar to FIG. 5 showing an alternative construction useful in the practice of the invention.

FIG. 7 is an exploded view of a modification of the hollow disk of FIG. 4 comprising a fifth embodiment of the invention.

FIG. 8 is an assembly view of the fifth embodiment of the invention.

FIG. 9 is an exploded view of a modification of the hollow disk of FIG. 4 comprising a sixth embodiment of the invention.

FIG. 10 is an assembly view of the sixth embodiment of the invention.

FIG. 11 is an exploded view of a modification of the hollow disk of FIG. 4 comprising a seventh embodiment of the invention.

FIG. 12 is an assembly view of the seventh embodiment of the invention.

FIG. 13 is an exploded view of a modification of the hollow disk of FIG. 4 comprising an eighth embodiment of the invention.

FIG. 14 is an assembly view of the eighth embodiment of the invention.

FIG. 15 is a diagrammatic illustration of a ninth embodiment of the invention.

FIG. 16 is a flow chart illustrating a tenth embodiment of the invention.

DETAILED DESCRIPTION

Referring now to the Drawings, and particularly to FIG. 1 thereof, there is shown an apparatus for manufacturing methanol **10** comprising a first embodiment of the invention. The apparatus **10** includes a gas permeable tube **12** positioned within a glass tube **14**. The tube **12** can comprise sintered stainless steel, or sintered glass, sintered ceramic materials, or a photocatalytic material. As illustrated in FIG. 1, both the gas permeable tube **12** and the glass tube **14** comprise right circular cylinders with the tube **12** extending concentrically relative to the tube **14**. Other geometrical configurations of and positional relationships between the gas permeable tube **12** and the glass tube **14** may be utilized in accordance with the requirements of particular applications of the invention.

If not formed from a photocatalytic material, the gas permeable tube **12** has a light-activated catalyst layer **16** formed on the exterior surface thereof. The catalyst layer **16** is preferably a titanium-based catalyst; however, it will be understood that any light-activated catalyst which forms hydroxyl radicals from water may be utilized in the practice of the invention, if desired. A plurality of electromagnetic radiation sources **18**, such as ultraviolet lamps, are positioned around the exterior of the glass tube **14**, it being understood that while only one source **18** is illustrated in FIG. 1, in actual practice a plurality of energy sources **18** are employed and are disposed around the entire periphery of the tube **14**. As illustrated by the waves **20** in FIG. 1, the sources **18** generate energy in the form of, for example, ultraviolet light which is directed through the glass tube **14** and onto the catalytic layer **16** formed on the exterior surface of the gas permeable tube **12**.

In the operation of the apparatus for manufacturing methanol **10**, a quantity of water is received in a reservoir **22**. Water from the reservoir **22** is directed into the annular space between the gas permeable tube **12** and the glass tube **14** through piping **24**. During the operation of the apparatus **10** water flows through the annulus between the gas permeable tube **12** and the glass tube **14** on a continuous basis.

A quantity of methane or natural gas is stored in a reservoir **26**. In the operation of the apparatus **10**, gas is directed from the reservoir **26** into the interior of the gas permeable tube **12** through piping **28**. The gas within the gas permeable tube **12** is maintained at a pressure high enough to cause the gas to pass through the walls of the tube **12** into the water and to prevent the flow of water into the interior of the tube **12**.

In the operation of the apparatus for manufacturing methanol **10**, the water flowing through the annular space between the gas permeable tube **12** and the glass tube **14** causes gas bubbles to be continuously stripped off the exterior surface of the tube **12**. In this manner the size of the gas bubbles is maintained in the sub-micron range. The sub-micron size of the gas bubbles provides an enormous methane surface area which in turn results in unprecedented reaction efficiency.

As the sub-micron size gas bubbles are produced by the flow of water over the exterior surface of the gas permeable tube **12**, energy from the sources **18** continuously engages the catalytic surface **16** formed on the exterior of the tube **12**. This generates hydroxyl radicals in the flowing water. It is theorized that the hydroxyl radicals homolytically cleave one or more of the carbon-hydrogen bonds in the methane thereby forming either molecules of hydrogen or molecules of water, depending upon the initiating radical, and methyl radicals. The methyl radicals combine either with the

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hydroxyl radicals to form methanol or with the hydrogen radicals to form methane.

Those skilled in the art will appreciate the fact that other chemical reactions are possible in the operation of the apparatus for manufacturing methanol **10**. For example there exists the possibility of a methyl-methyl radical reaction, and also the possibility of a hydrogen-hydrogen radical reaction. Both of these possibilities are extremely remote due to the relatively low concentrations of methyl radicals and hydrogen radicals at any given time.

It will be further understood that natural gas typically comprises up to 10% ethane and up to 2% propane in addition to methane. Therefore, if natural gas is used in the practice of the invention, the reaction products include ethanol, normal propanol, and isopropanol in addition to methanol.

The water flowing from the annulus between the gas permeable tube **12** and the glass tube **14** having the reaction products contained therein is directed to a distillation apparatus **30** through piping **32**. The distillation apparatus **30** separates the outflow from the space between the tube **12** and the tube **14** into at least four streams, including a stream of unreacted methane **34** which is returned to the reservoir **26**, a stream of water **36** which is returned to the reservoir **22**, a stream of other reaction products **38** which are recovered, and a stream of methanol **40**. The stream of other reaction products **38** may be further separated into its component parts, if desired.

The present invention further comprises a method of making methanol. In accordance with the method there is provided a continuously flowing stream of water. Sub-micron size bubbles of methane are continuously injected into the flowing water. Hydroxyl radicals are continuously generated from the water. It is theorized that the hydroxyl radicals cleave the hydrogen-carbon bonds of the methane to form methyl radicals. The methyl radicals combine with the hydroxyl radicals to form methanol.

In accordance with more specific aspects of the method, a gas permeable tube having a catalytic layer on the exterior surface thereof is positioned within a glass tube. Water is directed through the annulus between the gas permeable tube and the glass tube, and methane or natural gas is directed into the interior of the gas permeable tube. The water flowing between the gas permeable tube and the glass tube continuously strips sub-micron size bubbles from the exterior surface of the gas permeable tube.

Electromagnetic radiation from, for example, ultraviolet lamps is directed through the glass tube and engages the catalytic surface on the exterior of the gas permeable tube, thereby forming hydroxyl radicals from the flowing water. It is theorized that the hydroxyl radicals homolytically cleave one or more of the carbon-hydrogen bonds in the methane to form either molecules of hydrogen or molecules of water, and methyl radicals. The methyl radicals combine either with the hydroxyl radicals to form methanol or with the hydrogen radicals to form methane. Ethanol and propanol are also produced if natural gas is used in the process.

The use of an internal gas permeable partition cylinder is shown in FIG. 1. One skilled in the art would also recognize that a vast number of shapes and orientations could be used to accomplish the same purpose. For example, the glass tube **14** does not need to be shaped as a tube in order to be functional as a housing. In fact, such a housing need only be partially transparent to electromagnetic radiation for the apparatus to function. Additionally, the orientation of the gas inside an inner tube with water between the inner tube and

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a housing is not required. One skilled in the art could envision a housing bisected by a gas permeable partition creating a water chamber and a gas chamber. The only requirements of such an embodiment is that the water chamber has a water source and a product outlet, which leads to an isolation apparatus, preferably a distillation apparatus; the gas chamber has a gas source; the gas permeable partition has a catalytic layer that is exposed to electromagnetic energy on the water side of the partition; and the gas permeable partition allows the penetration of gas bubbles that are sheared off by the relative movement of water in the water chamber relative to the gas permeable membrane.

Referring now to FIG. 2, there is shown an apparatus for manufacturing methanol comprising a second embodiment of the invention. The apparatus **50** comprises numerous component parts which are substantially identical in construction and function to the apparatus for manufacturing methanol **10** shown in FIG. 1 and described hereinabove in conjunction therewith. Such identical component parts are designated in FIG. 2 with the same reference numerals utilized in the description of the apparatus **10**, but are differentiated therefrom by means of a prime (') designation.

In the apparatus for manufacturing methanol **50**, the gas permeable tube **12'** is supported for rotation relative to the glass tube **14'** by sealed bearings **52**. Those skilled in the art will appreciate the fact that bearing/seal assemblies comprising separate components may be utilized in the practice of the invention, if desired.

A motor **54** is mounted at one end of the glass tube **14'** and is operatively connected to the gas permeable tube **12'** to effect rotation thereof relative to the glass tube **14'**. The glass tube **14'** includes an end portion **56** which is isolated from the remainder thereof by a seal **58**. The portion of the tube **12'** extending into the end portion **56** of the glass tube **14'** is provided with a plurality of uniform or nonuniform apertures **60**.

In the operation of the apparatus for manufacturing methanol **50**, methane or natural gas is directed from the reservoir **26'** through the piping **28'**, through the end portion **56** of the glass tube **14'** and through the apertures **60** into the interior of the gas permeable tube **12'**. Water flows from the reservoir **22'** through the piping **24'** and into the portion of the glass tube **14'** that is isolated from the end portion **56** by the seal **58**. Water flows out of the glass tube **14'** through piping **32'** to the distillation apparatus **30'**.

The operation of the apparatus for manufacturing methanol **50** of FIG. 2 differs from the operation of the apparatus for manufacturing methanol **10** of FIG. 1 in that in the operation of the apparatus **50**, the relative movement between the bubbles forming on the surface of the gas permeable tube **12'** and the water contained within the glass tube **14'** is controlled by the motor **54** rather than the flow rate of the water as it passes through the glass tube **14'**. This is advantageous in that it allows the gas permeable tube **12'** to be rotated at a relatively high velocity relative to the water contained within the glass tube **14'**, thereby assuring that sub-micron size bubbles will be sheared from the surface of the gas permeable tube **12'**. Meanwhile, the velocity of the water passing through the interior of the glass tube **12'** can be relatively slow, thereby assuring a maximum number of sub-micron size bubbles entering the water per unit volume thereof.

An apparatus for manufacturing methanol comprising a third embodiment of the invention is illustrated in FIG. 3. The apparatus for manufacturing methanol **60** comprises

numerous component parts which are substantially identical in construction and function to component parts of the apparatus for manufacturing methanol **10** illustrated in FIG. **1** and described hereinabove in conjunction therewith. Such identical component parts are designated in FIG. **3** with the same reference numerals utilized in the description of the apparatus **10**, but are differentiated therefrom by means of a double prime (**41**) designation.

The apparatus for manufacturing methanol **60** comprises a gas permeable tube **12**" which is supported for rotation relative to the glass tube **14**" by sealed bearings **62**. Those skilled in the art will appreciate the fact that the apparatus **60** may be provided with bearing/seal assemblies comprising separate components, if desired.

The gas permeable tube **12**" is provided with one or more turbines **64**. The pitch of the turbines **64** is adjusted to cause the tube **12**" to rotate at a predetermined speed in response to a predetermined flow rate of water through the glass tube **14**".

Similarly to the apparatus for manufacturing methanol of FIG. **2**, the use of the apparatus for manufacturing methanol **60** is advantageous in that the gas permeable tube **12**" can be caused to rotate relatively rapidly in response to a relatively low flow rate of water through the glass tube **14**". This assures that sub-micron size bubbles will be stripped from the outer surface of the gas permeable tube **12**" and that a maximum number of bubbles will be received in the water flowing through the glass tube **14**" per unit volume thereof. The use of the apparatus for manufacturing methanol **60** is particularly advantageous in applications of the invention wherein water flows through the system under the action of gravity, in that the use of the turbines **64** eliminates the need for a separate power source to effect rotation of the gas permeable tube **12**" relative to the glass tube **14**".

Referring now to FIGS. **4** and **5**, there is shown a method of and apparatus for manufacturing methanol and other alcohols **70** comprising a fourth embodiment of the invention. In accordance with a fourth embodiment of the invention, there is provided a distillation unit **72** comprising a tank having a quantity of water **74** contained therein. One or more electromagnetic radiation sources **76** are also positioned in the tank **72**. The distillation unit **72** includes a heat source, which may comprise the radiation sources **76**, sufficient to effect distillation of methanol and other alcohols from water.

A hollow disk **78** is mounted in the lower portion of the tank **78**. As is best shown in FIG. **5**, the disk **78** includes a gas permeable partition **80** supported on a tube **82** for rotation within the tank **72** under the operation of a motor **84**. The partition **80** may comprise sintered stainless steel, sintered glass, or sintered ceramic materials, or may be formed entirely from a catalytic material, depending upon the requirements of particular applications of the invention. Natural gas received from a supply **86** is directed through piping **88** and a suitable commutator **90** into the tube **82** and through the tube **82** into the interior of the hollow disk **78**. The tube **82** has a hollow interior **90** and the disk **78** has a hollow interior **92** connected in fluid communication therewith. The gas permeable partition **80** is coated with a light-activated catalytic layer **94**.

The disk **78** is supplied with natural gas at a pressure just high enough to overcome to head pressure of the water **74**. The disk **78** is rotated by the motor **84** at an appropriate speed in contact with the water **74** such that a shearing phenomenon occurs at the surface of the photocatalytic layer **94** thus producing bubbles of natural gas of extremely small

size. The extreme small size of the bubbles thus produced results in a surface area to volume ratio of small bubbles which significantly improves the efficiency of the reaction.

As the sub-micron size gas bubbles are produced by movement of the exterior surface of the gas permeable partition **80** in the water **74**, electromagnetic energy from the sources **76** continuously engages the catalytic surface **94** formed on the exterior of the partition **80**, it being understood that depending on the characteristics of the catalytic layer **94**, energy comprising various portions of the electromagnetic spectrum may be used in the practice of the invention.

Activation of the catalytic layer **94** generates hydroxyl radicals in the water. It is theorized that the hydroxyl radicals homolytically cleave one or more of the carbon-hydrogen bonds in the methane, ethane, propane, etc., thereby forming either molecules of hydrogen or molecules of water, depending upon the initiating radical, and methyl, ethyl, and propyl radicals which combine either with the hydroxyl radicals to form methanol, ethanol, and propanol, or with the hydrogen radicals to form methane, ethane, and propane.

The methanol produced by the operation of the distillation unit **72** is recovered at outlet **96**. A pressure swing absorber **97** receives natural gas and hydrogen from the distillation unit **72**. Unreacted natural gas is recovered from the pressure swing absorber **97** at outlet **98** and is returned to the distillation unit **72** through piping **88**. Byproduct hydrogen produced in the distillation unit **72** is recovered at outlet **100** and is directed to a fuel cell **102**.

Within the fuel cell **102**, hydrogen recovered from the distillation unit **72** is combined with oxygen from the atmosphere to produce electricity which is recovered at terminal **104** and water which is recovered at outlet **106** and returned to the distillation unit **72** through piping **108**. As will be appreciated by those skilled in the art, a conventional engine/generator may be used in lieu of the fuel cell **102**; however, the use of a fuel cell is preferred due to its greater efficiency.

The use of the hydrogen recovered from the distillation unit **72** to produce electricity comprises an important advantage in the use of the present invention in that the electricity thus produced may be utilized to provide artificial lighting in those instances in which the apparatus **70** is situated at a remote location and/or to provide heating for the distillation unit **72**. As is shown in FIG. **4**, electricity from the fuel cell **102** may also be used to operate the radiation sources and **76** the motor **84**.

In addition to producing methanol, the apparatus **70** converts other alkanes present in the natural gas to their respective alcohols, namely: ethanol, normal propanol, and isopropanol. The higher alcohols thus produced are recovered from the distillation **72** at outlet **110** and are directed to a reverse osmosis unit **112**, and from the reverse osmosis unit **112** to a secondary distillation unit **114** to produce purer forms of the higher alcohols. Like the distillation unit **72**, the distillation unit **114** is provided with a heat source adequate to effect the desired distillation. Unrecovered materials from the secondary distillation unit **114** are returned to the reverse osmosis unit **112** through piping **116**. The reverse osmosis unit **112** also produces water which is returned to the distillation unit **72** through the piping **108**.

Typically, the water which is returned to the distillation unit **72** from the fuel cell **102** and the reverse osmosis unit **112** is sufficient to maintain a predetermined quantity of water therein. The distillation unit **72** is initially filled from a water supply **118** which is also available to supplement the

water received from the fuel cell **102** and the reverse osmosis unit **112** if necessary to maintain an adequate supply of water in the distillation unit **72**.

In lieu of the motor **84**, the disk **78** may be oscillated using a torsion motor or reciprocated using a motor and crank assembly. Other apparatus for effecting relative movement between the partition **80** and the water **74** will suggest themselves to those skilled in the art.

As will be appreciated by those skilled in the art, it is known to produce gas permeable partitions entirely from photocatalytic material, including titanium-based catalytic materials. FIG. 6 illustrates a hollow disk **78** having a gas permeable partition **120** formed entirely from one or more catalytic materials. Such construction eliminates the need of forming a catalytic layer on the surface of a gas permeable partition.

Those skilled in the art will appreciate the fact that the method and apparatus of the present invention can be utilized to convert gases other than methane and natural gas into valuable products. For example, the method and apparatus of the present invention can be utilized to convert carbon dioxide to methanol and methane. The adaptation of other chemical processes to the method and apparatus of the present invention will readily suggest themselves to those skilled in the art.

Referring to FIGS. 7 and 8, there is shown a hollow disk assembly **130** comprising a fifth embodiment of the invention. The hollow disk assembly **130** includes a hollow disk **132** which is supported on a hollow tube **134** for rotation, oscillation, or reciprocation relative to a quantity of water (not shown in FIGS. 7 and 8). The hollow disk assembly **130** further includes a porous partition **136** supported on the hollow disk **132** and a layer or plate of catalyst material **138** supported on the porous partition **136**.

The hollow disk assembly **130** comprising the fifth embodiment of the invention differs from the hollow disk assembly of FIG. 4 in two significant aspects. First, the hollow disk **132** is provided with a mirrored surface **140** formed on the interior surface of the hollow disk **132** opposite the porous partition **136**. Thus, the mirrored surface **140** functions to reflect electromagnetic radiation passing through the catalytic material **138** and the porous partition **136** back to the catalytic material **138**, thereby substantially increasing the efficiency of the interaction between the electromagnetic radiation and the catalytic material **138**.

Additionally, the porous plate **136** includes a quantity of a fluorescent material. The fluorescent material which is included in the porous partition **136** is selected to respond to broad-band electromagnetic radiation to produce an output comprising narrow band electromagnetic radiation which is specifically matched to the band width of the radiation which activates the catalytic material **138**. In this manner, the efficiency of the catalytic reaction is substantially increased because the portion of the electromagnetic radiation which would otherwise be unused is transformed by the fluorescent material into radiation within the band width comprising the input requirements of the catalytic material.

Referring to FIGS. 9 and 10, there is shown a hollow disk assembly **150** comprising a sixth embodiment of the invention. The hollow disk assembly **150** includes a hollow disk **152** which is supported on a hollow tube **154** for rotation, oscillation, or reciprocation relative to a quantity of water (not shown in FIGS. 9 and 10). The hollow disk assembly **150** further includes a porous partition **156** supported on the hollow disk **152** and a layer or plate of catalyst material **158** supported on the porous partition **156**.

The hollow disk assembly **150** comprising the sixth embodiment of the invention differs from the hollow disk assembly of FIG. 4 in two significant aspects. First, the hollow disk **152** is provided with a mirrored surface **160** formed on the interior surface of the hollow disk **152** opposite the porous partition **156**. Thus, the mirrored surface **160** functions to reflect electromagnetic radiation passing through the catalytic layer **158** and the porous partition **156** back to the catalytic layer **158**, thereby substantially increasing the efficiency of the interaction between the electromagnetic radiation and the catalytic layer.

Additionally, the catalyst material **158** includes a quantity of a fluorescent material. The fluorescent material which is included in the catalyst material **158** is selected to respond to broad-band electromagnetic radiation to produce an output comprising narrow band electromagnetic radiation which is specifically matched to the band width of the radiation which activates the catalytic material **158**. In this manner, the efficiency of the catalytic reaction is substantially increased because the portion of the electromagnetic radiation which would otherwise be unused is transformed by the fluorescent material into radiation within the band width comprising the input requirements of the catalytic layer.

Referring to FIGS. 11 and 12, there is shown a hollow disk assembly **170** comprising a seventh embodiment of the invention. The hollow disk assembly **170** includes a hollow disk **172** which is supported on a hollow tube **174** for rotation, oscillation, or reciprocation relative to a quantity of water (not shown in FIGS. 11 and 12). The hollow disk assembly **170** further includes a porous partition **176** supported on the hollow disk **172** and a layer or plate of catalyst material **178** supported on the porous partition **176**.

The hollow disk assembly **170** comprising the seventh embodiment of the invention differs from the hollow disk assembly of FIG. 4 in two significant aspects. First, the hollow disk **172** is provided with a mirrored surface **180** formed on the interior surface of the hollow disk **172** opposite the porous partition **176**. Thus, the mirrored surface **180** functions to reflect electromagnetic radiation passing through the catalytic layer **178** and the porous partition **176** back to the catalytic layer **178**, thereby substantially increasing the efficiency of the interaction between the electromagnetic radiation and the catalytic layer.

Second, in addition to the porous plate **176** and the layer or plate of catalyst material **178**, the seventh embodiment includes a plate **182** comprising a fluorescent material. The fluorescent material which is included in the plate **182** is selected to respond to broad-band electromagnetic radiation to produce an output comprising narrow band electromagnetic radiation which is specifically matched to the band width of the radiation which activates the catalytic material **178**. In this manner, the efficiency of the catalytic reaction is substantially increased because the portion of the electromagnetic radiation which would otherwise be unused is transformed by the fluorescent material into radiation within the band width comprising the input requirements of the catalytic layer.

In FIGS. 13 and 14 there is shown a hollow disk assembly **190** comprising an eighth embodiment of the invention. The hollow disk assembly **190** includes a hollow disk **192** which is supported on a hollow tube **194** for rotation, oscillation, or reciprocation relative to a quantity of water (not shown in FIGS. 13 and 14). A porous partition **196** is supported on the hollow disk **192** and in turn supports a plate or layer of catalytic material **198**.

The hollow disk assembly **190** differs from the hollow disk assemblies **130**, **150**, and **170** in that rather than employing a mirrored surface formed directly on the hollow disk **192**, there is provided a separate reflective disk **199**. The reflective disk **199** may be fabricated from glass or transparent plastic, in which case the interior surface thereof is provided with a reflective layer in the manner of a conventional mirror. Alternatively, the reflective disk **199** may comprise stainless steel or other metal having a highly polished exterior surface.

Those skilled in the art will appreciate the fact that a mirrored surface formed on an appropriate interior surface may be used in any of the embodiments of the invention illustrated in FIGS. **1** through **12**, inclusive, and described hereinabove in conjunction therewith. Likewise, a separate mirrored member or members having a variety of geometric configurations can be used in conjunction with any of the embodiments of the invention illustrated in FIGS. **1** through **12**, inclusive, and described hereinabove in conjunction therewith. Likewise, any of the fluorescent material constructions illustrated in FIGS. **7** through **12**, inclusive, and described hereinabove in conjunction therewith can be utilized in conjunction with any of the embodiments of the invention illustrated in FIGS. **1** through **6**, inclusive, **13** and **14** and described hereinabove in conjunction therewith.

Referring to FIG. **15**, there is shown a method of and apparatus for manufacturing methanol from methane or natural gas **200** comprising a ninth embodiment of the invention. In accordance with a ninth embodiment, a plurality of porous partitions **202** are mounted in a predetermined array which may be either linear, circular, three-dimensional, etc. The porous partition **102** may be tubular in shape, however, any desired geometrical configuration may be utilized in the construction of the porous partitions **102** depending upon the requirements of particular applications of the invention. Each of the porous partitions **202** has a photocatalytic layer formed on its exterior surface.

A partition **204** which is transparent to electromagnetic radiation is positioned on each side of each porous partition **202**. Within each transparent partition **204** there is provided a source of electromagnetic radiation **206** which may comprise, for example, a source of ultraviolet light, it being understood that other sources of electromagnetic radiation providing the same or different types of radiation may be utilized in the practice of the invention depending upon the requirements of particular applications thereof.

Methane or natural gas received from a source **208** is directed into the interior of each porous partition **202** from a first manifold **210**. Simultaneously, water received from a source **212** is directed through a second manifold **214** into the spaces between the porous partitions **202** and the electromagnetic radiation transparent partitions **204**. Within each porous partition **202** the pressure of the methane or natural gas is maintained just high enough to cause methane or natural gas to flow outwardly through the porous partition while preventing the flow of water inwardly through the porous partition.

In the operation of the apparatus **200**, relative movement is continuously effected between the water and the exterior surfaces of the porous partitions **202** using, for example, the techniques shown in FIGS. **1**, **2**, and **3** and described hereinabove in conjunction therewith. Electromagnetic radiation from the sources **206** activates the catalytic layers on the exterior surfaces of the porous partitions **202** to form hydroxyl radicals from the water. The hydroxyl radicals combine with the methane to form methanol, and if natural

gas is used in the operation of the apparatus **200**, to form methanol and higher alcohols. The energy from the electromagnetic radiation sources **206** is sufficient to distill the methanol, and, if present, the higher alcohols from the water for recovery at an outlet **216**. The remaining water is recovered at an outlet **218** and is returned to the manifold **214** for reuse.

A method of manufacturing methanol and higher alcohols from natural gas is shown in the flow chart comprising FIG. **16** which depicts a tenth embodiment of the invention. In accordance with the tenth embodiment, natural gas or methane is mixed with an oxidizer such as oxygen, peroxide, etc. in accordance with a predetermined ratio. The mixture comprising natural gas or methane and an oxidizer is then directed through a porous partition having a photocatalytic exterior surface. The photocatalytic exterior surface is surrounded by a quantity of water, and relative movement is continuously maintained between the photocatalytic surface and the water. Electromagnetic energy is directed onto the photocatalytic layer which forms hydroxyl radicals from the water. The hydroxyl radicals combine with methyl radicals from the methane to form methanol. If natural gas is used, the hydroxyl radicals combine with methyl, ethyl, and propyl radicals to form methanol and higher alcohols. The alcohol(s) thus produced are recovered along with byproduct hydrogen. The presence of the oxidizer in the mixture causes the reaction to be self-sustaining.

Although preferred embodiments of the invention have been illustrated in the accompanying Drawing and described in the foregoing Detailed Description, it will be understood that the invention is not limited to the embodiments disclosed but is capable of numerous rearrangements, modifications, and substitutions of parts and elements without departing from the spirit of the invention.

We claim:

1. A method of manufacturing methanol and higher alcohols from natural gas comprising the steps of:
 - providing a hollow chamber having a gas permeable partition;
 - providing a quantity of light activated catalytic material comprising at least the exterior of the gas permeable partition;
 - directing electromagnetic radiation onto the catalytic material, at least a portion of the electromagnetic radiation passing through the catalytic material and the gas permeable partition into the hollow chamber; and reflecting electromagnetic radiation passing through the porous partition back through the gas permeable partition and into engagement with the catalytic material.
2. The method of claim **1** further including the step of providing a reflective surface formed directly on part of the hollow chamber.
3. The method of claim **1** further including the step of providing a separate reflective member positioned within the hollow chamber.
4. The method of claim **1** further characterized by:
 - providing catalytic material characterized by a predetermined activation band width;
 - directing electromagnetic radiation having a band width encompassing but wider than the activation band width of the catalytic material onto the catalytic material;
 - providing a quantity of fluorescent material responsive to electromagnetic radiation for producing an output radiation having a band width at least partially within the activation band of the catalytic material; and directing the output radiation from the fluorescent material into engagement with the catalytic material.

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5. The method of claim 4 further including the step of incorporating the fluorescent material into the gas permeable partition of the hollow chamber.

6. The method of claim 4 further including the steps of: providing a layer of catalytic material separate from the gas permeable partition; and incorporating the fluorescent material into the catalytic material.

7. The method of claim 4 further including the steps of: providing a layer of catalytic material separate from the gas permeable partition; providing a layer of fluorescent material separate from the layer of catalytic material and the gas permeable partition; and

positioning the layer of fluorescent material between the layer of catalytic material and the gas permeable partition.

8. A method of manufacturing methanol and higher alcohols from natural gas comprising the steps of:

providing a hollow chamber including a gas permeable partition;

providing a quantity of light-activated catalytic material having a predetermined activation band width comprising at least the exterior of the gas permeable partition;

directing electromagnetic radiation having a predetermined band width wider than and encompassing the activation band width of the catalytic material onto the catalytic material;

providing a quantity of fluorescent material responsive to the electromagnetic radiation from the source for producing output radiation having a band width at least partially within the activation band width of the catalytic material; and

directing the output radiation from the fluorescent material into engagement with the catalytic material.

9. The method of claim 8 further including the steps of incorporating the fluorescent material into the gas permeable partition of the hollow chamber.

10. The method of claim 8 including the additional steps of providing a layer of catalytic material separate from the gas permeable partition and incorporating the fluorescent material into the layer of catalytic material.

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11. The method of claim 8 wherein at least part of the electromagnetic radiation from the source passes through the catalytic material and the gas permeable partition into the hollow chamber and including the additional step of providing the reflective surface within the hollow chamber for reflecting electromagnetic radiation passing through the gas permeable partition back through the gas permeable partition and into engagement with the fluorescent material and the catalytic material.

12. The method of claim 11 including the additional step of forming the reflective surface on part of the hollow chamber.

13. The improvement according to claim 12 including additional steps for providing a separate member positioned within the hollow chamber.

14. A method for manufacturing methanol and higher alcohols from natural gas including the steps of:

providing a hollow chamber including a gas permeable partition;

providing a quantity of light-activated catalytic material having a predetermined activation band width comprising at least the exterior of the gas permeable partition;

directing electromagnetic radiation having a predetermined band width encompassing the activation band width of the catalytic material onto the catalytic material, at least a portion of the electromagnetic radiation passing through the catalytic material and the gas permeable partition into the hollow chamber;

providing the quantity of fluorescent material responsive to said electromagnetic radiation from the source for producing output radiation having a band width at least partially within the activation band width of the catalytic material;

directing the output radiation from the fluorescent material into engagement with the catalytic material; and

providing a reflective surface within the hollow chamber for reflecting electromagnetic radiation passing through the gas permeable partition back through the gas permeable partition and into engagement with the catalytic material and the fluorescent material.

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